

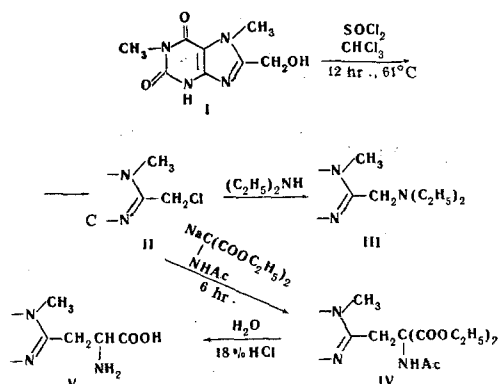
THE SYNTHESIS OF SOME NEW DERIVATIVES OF 1,7-DIMETHYLXANTHINE

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We have synthesized a group of new 8-substituted 1,7-dimethylxanthines which are of interest for biological investigation as very close structural analogs of some 8-substituted theobromine derivatives obtained previously. In particular, we have carried out the conversion of 8-hydroxymethyl-1,7-dimethylxanthine (I) [1] into 8-chloromethyl-1,7-dimethylxanthine (II) and from this, by heating it for 6 hours with an excess of diethylamine or by condensing it with sodioacetylaminomalonic ester in ethanol we have obtained 8-diethylaminomethyl-1,7-dimethylxanthine (III) and its hydrochloride (compare [2]) or 8-(β -acetyl-amino- β , β -diethoxycarbonyl-ethyl)-1,7-dimethylxanthine (IV). The hydrolytic cleavage of the latter, accompanied by partial decarboxylation, led to 1,7-dimethyl-8-xanthinylalanine (V) (cf. [3]).



II: mp 250°–251° C (decomp., from a mixture of ethanol and dimethylformamide.) Found, %: Cl 15.47; N 24.63. Calculated for $C_8H_9ClN_4O_2$, %: Cl 15.53; N 24.57.

III: mp 192°–194° C (from 30% ethanol). Found, %: N 25.77. Calculated for $C_{12}H_{19}N_5O_2$, %: N 26.41. Hydrochloride: mp 260°–262° C (decomp.). Found, %: Cl 11.57; N 23.00. Calculated for $C_{12}H_{19}N_5O_2 \cdot HCl$, %: Cl 11.77; N 23.22.

IV: mp 229°–230° C (from ethanol). Found, %: C 50.17; H 5.79; N 17.29. Calculated for $C_{17}H_{23}N_5O_7$, %: C 49.87; H 5.62; N 17.11.

V: mp 265° C (from water). Found, %: C 42.40; H 5.13; N 24.56; H_2O 6.0. Calculated for $C_{10}H_{13}N_5O_4 \cdot H_2O$, %: C 42.10; H 5.20; N 24.55; H_2O 6.3.

REFERENCES

- V. B. Kalcheva, E. S. Golovchinskaya, KhGS [Chemistry of Heterocyclic Compounds], (in press).
- E. S. Golovchinskaya, Mokhammed Yasin Ebed, and E. S. Chaman, ZhOkh, 32, 4098, 1962.
- E. S. Chaman, A. A. Cherkasova, and E. S. Golovchinskaya, ZhOkh, 30, 1878, 1960.

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SOME REACTIONS OF PHOSPHORUS-CONTAINING HETEROCYCLES

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It has been reported previously that derivatives of dioxaphospholene [1] and oxaphospholene [2] react readily with water. The reaction, which is a saponification of a cyclic ester, ends with the formation of the corresponding acid, which contains a keto group in the organic

part of the molecule. The reaction of the cyclic compounds mentioned with alcohols has not been studied.

We have found [3] that derivatives of oxaphospholene readily react with alcohols on being heated, forming the corresponding esters of

Esters of Ketophosphonic and Ketophosphinic Acids

Compound	R	R'	Bp, °C (pressure, mm)	n_D^{20}	d_4^{20}	Empirical formula	Found				Calculated				Yield, %
							C, %	H, %	P, %	MR _D	C, %	H, %	P, %	MR _D	
I	OCH ₃	C ₂ H ₅	73–74 (0.04)	1.4467	1.0896	C ₉ H ₁₀ O ₄ P	48.67	8.43	13.81	54.45	48.64	8.62	13.94	54.64	65.8
II	OC ₂ H ₅ -n	C ₂ H ₅	76–78 (0.02)	1.4429	1.0235	C ₁₂ H ₁₆ O ₄ P	54.46	10.03	11.68	68.46	54.53	9.53	11.72	68.49	41.6
III	C ₂ H ₅	CH ₃	71–72 (0.1)	1.4540	1.0638	C ₉ H ₁₁ O ₄ P	52.62	9.39	15.38	53.48	52.42	9.29	15.02	53.52	71.6
IV	C ₂ H ₅	C ₂ H ₅	78–80 (0.07)	1.4595	1.0362	C ₁₀ H ₁₂ O ₄ P	54.48	9.67	13.91	58.15	54.53	9.61	14.06	58.14	30.8
V	C ₂ H ₅	H	MP, 112–113	—	—	C ₈ H ₁₇ O ₄ P	49.52	8.64	15.78	192.10*	50.00	8.92	16.11	192.18*	89.0

*Equivalent weight, found 192.10, calculated 192.18.